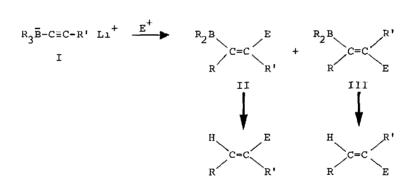
A NEW ROUTE TO DI- AND TRISUBSTITUTED OLEFINS VIA LITHIUM TRIALKYLALKYNYLBORATES

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The rearrangement of trialkylalkynylborate salts (I) has been induced by a variety of electrophiles, and the stereochemical outcome is a complex function of several factors, including the nature of the electrophile, the type of organoborane, and the solvent.¹⁻⁵ However, direct protonation of I invariably produces a mixture of <u>cis</u>- and <u>trans</u>-vinylboranes (II and III), which leads ultimately to <u>cis</u>-trans olefin mixtures (E = H).^{1,6,7}



We have investigated the reactions of I with several electrophiles (R_n^MX) of Groups III, IV, and VI with a view of forming specifically-substituted vinyl intermediates (II or III, $E = MR_n$), and determining if the boron-carbon and "metal"-carbon bond can be exploited separately.

Reaction of 1-lithiohexyne (from 5 mmol 1-hexyne and 5 mmol $\underline{n}-C_4H_9Li$ (Alfa, hexane)) in 7 ml THF with 5 mmol Et₃B (Alfa) at 0° for 1 hour produced borate I (R = Et, R' = $\underline{n}-C_4H_9$). Then tributyltin chloride (5 mmol, Alfa) was added at room temperature, and the mixture stirred for 1 hour. The precipitate of LiCl was filtered, and solvent removed <u>in vacuo</u>. Mass spectral analysis of the remaining pale amber residue indicated a parent ion, m/e 470 (¹¹B, ¹²⁰Sn). Oxidation (3N NaOAc, 30% H₂O₂) of the original crude mixture produced 3-octanone (88%), and hydrolysis (<u>formic acid</u>) produced a <u>single</u> olefin shown (vpc, ¹H nmr) to be <u>cis</u>-3-octene (82%). Other hydrolytic media - including HOAc, $EtCO_2H$, pivalic acid, H₂O, aq. NH₄Cl, HCl, CH₃OH and thiophenol - proved less efficient and produced varying amounts (up to 15%) of 3-ethyl-3-octene, the product of "double migration".^{8,9}

Thus, the reaction of borates I with Bu_3SnCl specifically produces the isomer in which the migrating group and entering electrophile are <u>trans</u> (II, $E = SnBu_3$). Yields of other olefins (5 mmol scale, HCO_2H hydrolysis) are shown in the Table.

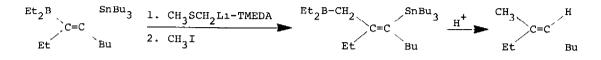
Table

$$R_3B$$
 $R'CECH$ Olefin (a)Yield (b) Et_3B hexyne cis -3-octene82 Bu_3B hexyne cis -5-decene79 Et_3B heptyne cis -3-nonene77 (71) Et_3B phenylacetylene cis -1-phenyl-1-butene73 Hex_3B hexyne cis -5-dodecene89 (70) $\bigvee_{0} (CH_2)_{4} \xrightarrow{1}_{3} B$ hexyne $(CH_2)_{4} \xrightarrow{-C=C-n-C_4}H_9$ (45)

(a) satisfactory ir, pmr, mass spectra and/or vpc comparison with known olefins.(b) yields in parentheses are isolated, pure material.

Presumably the bulk of the tributyl tin molety restricts the olefin synthesis to the use of less branched borates. For example, no olefin was observed with disiamylheptylhexynyl borate, disiamyl-l-hexenylhexynyl borate, (tris)-2methylpentylhexynyl borate, tricyclohexylhexynyl borate or triethyl-t-butylethynyl borate. Dicyclohexylheptylhexynyl borate gave a mixture of two olefins (50%), 1-cyclohexyl-1-hexene and 5-tridecene, in a 1 3 ratio respectively.

The intermediate mixed vinylorganometallic also serves as a precursor to tri-substituted olefins. Reaction of II (R = Et, R' = $\underline{n}-C_4H_9$, E = SnBu₃) with thiomethoxymethyllithium-TMEDA¹⁰ (2 equiv dimethyl sulfide, l equiv $\underline{n}-C_4H_9Li$. TMEDA), produced Z-3-methyl-3-octene (52%).



In contrast to Bu_3SnCl , electrophiles such as Et_2AlCl and C_6H_5SCl react with salts I (R = Et, R' = $\underline{n}-C_4H_9$) in a non-stereospecific fashion, producing intermediates (presumably II and III, E = $AlEt_2$ and SC_6H_5 , respectively) which form <u>cis-trans</u> mixtures on hydrolysis.

Interestingly, reaction of I with C₆H₅SeCl followed by acid hydrolysis gives rise exclusively to vinylselenide IV. The utility of these reagents will be reported separately.

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